Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

2-Isobutyl-6-(4-methoxyphenyl)-imidazo[2,1-*b*][1,3,4]thiadiazole

Hoong-Kun Fun,[a]*‡ Chin Sing Yeap,[a]§ D. Jagadeesh Prasad,[b] Prakash Anil Castelino[c] and V. V. Anitha[b]

[a]X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, [b]Department of Chemistry, Mangalore University, Mangalore, Karnataka, India, and [c]St. Philomena's College, Puttur, Dakshina Kannada, Karnataka, India
Correspondence e-mail: hkfun@usm.my

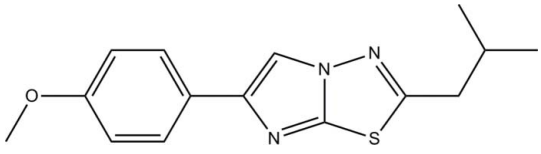
Received 17 December 2010; accepted 18 December 2010

Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.002 Å; R factor = 0.053; wR factor = 0.163; data-to-parameter ratio = 34.3.

In the title compound, $C_{15}H_{17}N_3OS$, the dihedral angle between the statistically planar imidazo[2,1-*b*][1,3,4]thiadiazole fused-ring system (r.m.s. deviation = 0.002 Å) and the methyoxbenzene ring is 4.52 (6)°. In the crystal, molecules are arranged into columns and stacked down the *a* axis. The crystal structure is stabilized by weak C—H$\cdots\pi$ and π–π interactions [centroid–centroid separations = 3.6053 (8) and 3.7088 (7) Å].

Related literature

For a related structure and background references to imidazo-[2,1-*b*]-1,3,4-thiadiazole derivatives, see: Fun *et al.* (2011).



Experimental

Crystal data

$C_{15}H_{17}N_3OS$ M_r = 287.38
Triclinic, $P\overline{1}$
a = 5.7139 (1) Å
b = 10.1795 (1) Å
c = 12.9689 (2) Å
α = 85.174 (1)°
β = 85.164 (1)°
γ = 80.690 (1)°
V = 739.84 (2) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 0.22 mm^{-1}
T = 296 K
0.43 × 0.31 × 0.17 mm

Data collection

Bruker SMART APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)
T_{min} = 0.911, T_{max} = 0.964
23503 measured reflections
6213 independent reflections
3805 reflections with $I > 2\sigma(I)$
R_{int} = 0.028

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.053
$wR(F^2)$ = 0.163
S = 1.03
6213 reflections
181 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.30 e Å^{-3}
$\Delta\rho_{min}$ = −0.28 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

*Cg*3 is the centroid of the C1–C6 benzene ring.

D—H$\cdots A$	D—H	H$\cdots A$	$D\cdots A$	D—H$\cdots A$
C11—H11*A*$\cdots$*Cg*3^{i}	0.97	2.60	3.5063 (16)	155

Symmetry code: (i) $-x+2, -y+2, -z+1$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

HKF and CSY thank Universiti Sains Malaysia (USM) for the Research University Grant No. 1001/PFIZIK/811160.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5778).

References

Bruker (2009). *APEX2*, *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
Fun, H.-K., Hemamalini, M., Prasad, D. J., Castelino, P. A. & Anitha, V. V. (2011). Acta Cryst. E67, o254.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Spek, A. L. (2009). *Acta Cryst.* D**65**, 148–155.

‡ Thomson Reuters ResearcherID: A-3561-2009.
§ Thomson Reuters ResearcherID: A-5523-2009.

supplementary materials

Acta Cryst. (2011). E**67**, o255 [**doi:10.1107/S1600536810053225**]

2-Isobutyl-6-(4-methoxyphenyl)imidazo[2,1-*b*][1,3,4]thiadiazole

H.-K. Fun, C. S. Yeap, D. J. Prasad, P. A. Castelino and V. V. Anitha

Comment

As part of our ongoing synthetic and structural studies of imidazo[2,1-*b*]-1,3,4-thiadiazole derivatives (Fun *et al.*, 2011), we now report the structure of the title compound, (I).

The mean plane through the imidazo[2,1-*b*]-1,3,4-thiadiazole ring and the methoxyphenyl moiety is essentially planar with the maximum deviation of 0.045 Å for atom C2 (Fig. 1). The isobutyl is twisted away from this mean plane with torsion angles of C9–C11–C12–C13 = 64.3 (2)° and C9–C11–C12–C14 = -172.81 (17)°. In the crystal structure, the molecules are arranged into columns and stacked down *a* axis (Fig. 2). The molecules are stabilized by the weak $Cg1\cdots Cg2^{i}$ = 3.7088 (7) Å, $Cg2\cdots Cg2^{i}$ = 3.6053 (8) Å and C11–H11A$\cdots Cg3^{i}$ interactions [*Cg*1, *Cg*2 and *Cg*3 are centroids of S1/C9/N1/N2/C10, N2/C8/C7/N3/C10 and C–C6 ring respectively; (i) 2 - *x*, 2 - *y*, 1 - *z*].

Experimental

5-Isobutyl-1,3,4-thiadiazol-2-amine (1 molar equivalent) and 4-methoxyphenacylbromide (1 molar equivalent) are refluxed with ethanol for 4 h. The solvent was then distilled and the reaction mass was poured onto the crushed ice. The resulting solid that separated out was filtered and dried. The compound was re-crystallized using ethanol and DMF mixture to yield colourless blocks of (I). M.P.: 118–122°C.

Refinement

All hydrogen atoms were positioned geometrically [C–H = 0.93–0.98 Å] and refined using a riding model, with U_{iso}(H) = 1.2 or 1.5U_{eq}(C).

Figures

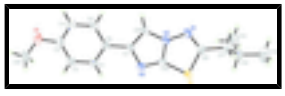


Fig. 1. The molecular structure of (I) with 30% probability ellipsoids for non-H atoms.

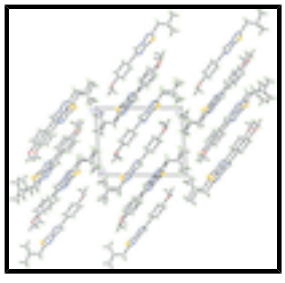


Fig. 2. The crystal packing of (I), viewed down the *a* axis, showing molecules stacked down *a* axis.

2-Isobutyl-6-(4-methoxyphenyl)imidazo[2,1-*b*][1,3,4]thiadiazole

Crystal data

$C_{15}H_{17}N_3OS$	$Z = 2$
$M_r = 287.38$	$F(000) = 304$
Triclinic, $P\overline{1}$	$D_x = 1.290$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 5.7139$ (1) Å	Cell parameters from 6150 reflections
$b = 10.1795$ (1) Å	$\theta = 2.5$–$30.1°$
$c = 12.9689$ (2) Å	$\mu = 0.22$ mm^{-1}
$\alpha = 85.174$ (1)°	$T = 296$ K
$\beta = 85.164$ (1)°	Block, colourless
$\gamma = 80.690$ (1)°	0.43 × 0.31 × 0.17 mm
$V = 739.84$ (2) Å^3	

Data collection

Bruker SMART APEXII CCD diffractometer	6213 independent reflections
Radiation source: fine-focus sealed tube	3805 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.028$
φ and ω scans	$\theta_{max} = 34.5°$, $\theta_{min} = 1.6°$
Absorption correction: multi-scan (*SADABS*; Bruker, 2009)	$h = -9 \rightarrow 9$
$T_{min} = 0.911$, $T_{max} = 0.964$	$k = -16 \rightarrow 16$
23503 measured reflections	$l = -20 \rightarrow 20$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.053$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.163$	H-atom parameters constrained
$S = 1.03$	$w = 1/[\sigma^2(F_o^2) + (0.0765P)^2 + 0.0748P]$ where $P = (F_o^2 + 2F_c^2)/3$
6213 reflections	$(\Delta/\sigma)_{max} < 0.001$
181 parameters	$\Delta\rho_{max} = 0.30$ e Å^{-3}
0 restraints	$\Delta\rho_{min} = -0.28$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations

between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
S1	1.14439 (6)	1.02701 (4)	0.26563 (3)	0.05911 (14)
O1	1.1085 (2)	0.33486 (12)	0.78317 (9)	0.0741 (3)
N1	0.7031 (2)	1.05243 (12)	0.33719 (9)	0.0523 (3)
N2	0.82991 (18)	0.94641 (11)	0.39100 (8)	0.0448 (2)
N3	1.16801 (18)	0.81311 (11)	0.42311 (8)	0.0470 (2)
C1	1.2380 (2)	0.58244 (13)	0.57132 (10)	0.0459 (3)
H1A	1.3637	0.6030	0.5258	0.055*
C2	1.2769 (3)	0.47212 (13)	0.64264 (10)	0.0498 (3)
H2A	1.4261	0.4199	0.6442	0.060*
C3	1.0917 (3)	0.44125 (14)	0.71078 (10)	0.0515 (3)
C4	0.8704 (3)	0.52078 (17)	0.70788 (12)	0.0620 (4)
H4A	0.7456	0.5008	0.7542	0.074*
C5	0.8345 (2)	0.62946 (16)	0.63658 (12)	0.0541 (3)
H5A	0.6852	0.6817	0.6354	0.065*
C6	1.0186 (2)	0.66223 (12)	0.56621 (9)	0.0405 (2)
C7	0.9808 (2)	0.77580 (12)	0.48936 (9)	0.0397 (2)
C8	0.7710 (2)	0.85747 (13)	0.47028 (10)	0.0497 (3)
H8A	0.6217	0.8533	0.5038	0.060*
C9	0.8461 (2)	1.10357 (13)	0.26920 (10)	0.0473 (3)
C10	1.0665 (2)	0.91576 (13)	0.36590 (9)	0.0441 (3)
C11	0.7678 (3)	1.22647 (14)	0.20133 (11)	0.0571 (4)
H11A	0.8370	1.2992	0.2240	0.069*
H11B	0.5965	1.2493	0.2123	0.069*
C12	0.8305 (3)	1.21810 (16)	0.08633 (11)	0.0642 (4)
H12A	1.0026	1.1905	0.0754	0.077*
C13	0.7083 (6)	1.1178 (2)	0.04179 (17)	0.1097 (9)
H13A	0.7482	1.0322	0.0783	0.165*
H13B	0.7594	1.1119	−0.0303	0.165*
H13C	0.5393	1.1455	0.0489	0.165*
C14	0.7663 (6)	1.3557 (2)	0.03141 (17)	0.1117 (9)
H14A	0.8461	1.4183	0.0607	0.167*
H14B	0.5975	1.3836	0.0400	0.167*
H14C	0.8148	1.3522	−0.0411	0.167*
C15	1.3354 (4)	0.25718 (19)	0.79377 (16)	0.0840 (6)
H15A	1.3233	0.1863	0.8467	0.126*
H15B	1.4436	0.3125	0.8127	0.126*
H15C	1.3931	0.2200	0.7291	0.126*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.04595 (19)	0.0694 (3)	0.0549 (2)	−0.00092 (16)	0.00022 (14)	0.01745 (17)
O1	0.0866 (8)	0.0673 (7)	0.0650 (7)	−0.0162 (6)	−0.0051 (6)	0.0229 (5)
N1	0.0439 (6)	0.0567 (7)	0.0504 (6)	0.0074 (5)	−0.0069 (5)	0.0047 (5)
N2	0.0364 (5)	0.0504 (6)	0.0438 (5)	0.0022 (4)	−0.0044 (4)	0.0022 (4)
N3	0.0361 (5)	0.0551 (6)	0.0462 (6)	−0.0009 (4)	−0.0031 (4)	0.0060 (4)
C1	0.0453 (6)	0.0489 (7)	0.0394 (6)	0.0004 (5)	0.0018 (5)	0.0001 (5)
C2	0.0532 (7)	0.0483 (7)	0.0440 (6)	0.0019 (5)	−0.0021 (5)	−0.0004 (5)
C3	0.0644 (8)	0.0468 (7)	0.0441 (6)	−0.0126 (6)	−0.0077 (6)	0.0034 (5)
C4	0.0531 (8)	0.0731 (10)	0.0584 (8)	−0.0177 (7)	0.0027 (6)	0.0119 (7)
C5	0.0394 (6)	0.0654 (8)	0.0546 (7)	−0.0063 (6)	0.0001 (5)	0.0060 (6)
C6	0.0395 (6)	0.0448 (6)	0.0374 (5)	−0.0054 (5)	−0.0040 (4)	−0.0050 (4)
C7	0.0372 (5)	0.0445 (6)	0.0365 (5)	−0.0023 (4)	−0.0035 (4)	−0.0050 (4)
C8	0.0375 (6)	0.0565 (8)	0.0505 (7)	−0.0008 (5)	0.0019 (5)	0.0051 (6)
C9	0.0514 (7)	0.0489 (7)	0.0394 (6)	0.0027 (5)	−0.0096 (5)	−0.0042 (5)
C10	0.0364 (6)	0.0528 (7)	0.0409 (6)	−0.0022 (5)	−0.0038 (4)	0.0010 (5)
C11	0.0711 (9)	0.0495 (7)	0.0465 (7)	0.0050 (6)	−0.0117 (6)	0.0006 (5)
C12	0.0702 (10)	0.0673 (9)	0.0486 (8)	0.0017 (8)	−0.0023 (7)	0.0078 (7)
C13	0.175 (3)	0.0923 (15)	0.0672 (12)	−0.0145 (17)	−0.0461 (15)	−0.0119 (11)
C14	0.164 (3)	0.0872 (14)	0.0717 (13)	−0.0059 (16)	−0.0039 (14)	0.0319 (11)
C15	0.1056 (16)	0.0615 (10)	0.0816 (12)	−0.0074 (10)	−0.0270 (11)	0.0232 (9)

Geometric parameters (Å, °)

S1—C10	1.7290 (13)	C6—C7	1.4622 (16)
S1—C9	1.7545 (14)	C7—C8	1.3722 (17)
O1—C3	1.3690 (16)	C8—H8A	0.9300
O1—C15	1.416 (2)	C9—C11	1.4966 (18)
N1—C9	1.2864 (19)	C11—C12	1.510 (2)
N1—N2	1.3721 (14)	C11—H11A	0.9700
N2—C10	1.3554 (16)	C11—H11B	0.9700
N2—C8	1.3689 (16)	C12—C13	1.503 (3)
N3—C10	1.3133 (15)	C12—C14	1.521 (2)
N3—C7	1.3957 (16)	C12—H12A	0.9800
C1—C6	1.3825 (17)	C13—H13A	0.9600
C1—C2	1.3947 (17)	C13—H13B	0.9600
C1—H1A	0.9300	C13—H13C	0.9600
C2—C3	1.379 (2)	C14—H14A	0.9600
C2—H2A	0.9300	C14—H14B	0.9600
C3—C4	1.388 (2)	C14—H14C	0.9600
C4—C5	1.381 (2)	C15—H15A	0.9600
C4—H4A	0.9300	C15—H15B	0.9600
C5—C6	1.3967 (18)	C15—H15C	0.9600
C5—H5A	0.9300		
C10—S1—C9	88.40 (6)	N3—C10—N2	112.76 (11)

C3—O1—C15	117.66 (14)	N3—C10—S1	138.85 (10)
C9—N1—N2	108.55 (11)	N2—C10—S1	108.39 (9)
C10—N2—C8	107.69 (10)	C9—C11—C12	116.36 (12)
C10—N2—N1	118.51 (11)	C9—C11—H11A	108.2
C8—N2—N1	133.79 (11)	C12—C11—H11A	108.2
C10—N3—C7	103.73 (10)	C9—C11—H11B	108.2
C6—C1—C2	122.09 (12)	C12—C11—H11B	108.2
C6—C1—H1A	119.0	H11A—C11—H11B	107.4
C2—C1—H1A	119.0	C13—C12—C11	112.03 (16)
C3—C2—C1	119.33 (13)	C13—C12—C14	110.67 (17)
C3—C2—H2A	120.3	C11—C12—C14	109.10 (15)
C1—C2—H2A	120.3	C13—C12—H12A	108.3
O1—C3—C2	124.39 (14)	C11—C12—H12A	108.3
O1—C3—C4	115.99 (14)	C14—C12—H12A	108.3
C2—C3—C4	119.62 (13)	C12—C13—H13A	109.5
C5—C4—C3	120.36 (14)	C12—C13—H13B	109.5
C5—C4—H4A	119.8	H13A—C13—H13B	109.5
C3—C4—H4A	119.8	C12—C13—H13C	109.5
C4—C5—C6	121.14 (14)	H13A—C13—H13C	109.5
C4—C5—H5A	119.4	H13B—C13—H13C	109.5
C6—C5—H5A	119.4	C12—C14—H14A	109.5
C1—C6—C5	117.46 (12)	C12—C14—H14B	109.5
C1—C6—C7	121.12 (11)	H14A—C14—H14B	109.5
C5—C6—C7	121.42 (11)	C12—C14—H14C	109.5
C8—C7—N3	111.03 (11)	H14A—C14—H14C	109.5
C8—C7—C6	127.62 (11)	H14B—C14—H14C	109.5
N3—C7—C6	121.35 (10)	O1—C15—H15A	109.5
N2—C8—C7	104.79 (11)	O1—C15—H15B	109.5
N2—C8—H8A	127.6	H15A—C15—H15B	109.5
C7—C8—H8A	127.6	O1—C15—H15C	109.5
N1—C9—C11	122.38 (13)	H15A—C15—H15C	109.5
N1—C9—S1	116.15 (10)	H15B—C15—H15C	109.5
C11—C9—S1	121.36 (11)		
C9—N1—N2—C10	0.23 (17)	C10—N2—C8—C7	0.06 (15)
C9—N1—N2—C8	179.94 (14)	N1—N2—C8—C7	−179.67 (13)
C6—C1—C2—C3	−0.3 (2)	N3—C7—C8—N2	0.03 (15)
C15—O1—C3—C2	4.8 (2)	C6—C7—C8—N2	−179.34 (11)
C15—O1—C3—C4	−175.43 (15)	N2—N1—C9—C11	−176.23 (11)
C1—C2—C3—O1	179.30 (13)	N2—N1—C9—S1	−0.04 (15)
C1—C2—C3—C4	−0.4 (2)	C10—S1—C9—N1	−0.10 (12)
O1—C3—C4—C5	−179.11 (14)	C10—S1—C9—C11	176.13 (11)
C2—C3—C4—C5	0.7 (2)	C7—N3—C10—N2	0.14 (15)
C3—C4—C5—C6	−0.2 (2)	C7—N3—C10—S1	−179.92 (13)
C2—C1—C6—C5	0.7 (2)	C8—N2—C10—N3	−0.13 (16)
C2—C1—C6—C7	−178.73 (12)	N1—N2—C10—N3	179.65 (11)
C4—C5—C6—C1	−0.5 (2)	C8—N2—C10—S1	179.92 (9)
C4—C5—C6—C7	178.95 (13)	N1—N2—C10—S1	−0.31 (15)
C10—N3—C7—C8	−0.10 (14)	C9—S1—C10—N3	−179.73 (16)
C10—N3—C7—C6	179.31 (11)	C9—S1—C10—N2	0.21 (10)

C1—C6—C7—C8	175.00 (13)	N1—C9—C11—C12	−129.32 (16)
C5—C6—C7—C8	−4.5 (2)	S1—C9—C11—C12	54.69 (18)
C1—C6—C7—N3	−4.31 (18)	C9—C11—C12—C13	64.3 (2)
C5—C6—C7—N3	176.24 (12)	C9—C11—C12—C14	−172.81 (17)

Hydrogen-bond geometry (Å, °)

Cg3 is the centroid of the C1–C6 benzene ring.

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C11—H11A···Cg3^{i}	0.97	2.60	3.5063 (16)	155

Symmetry codes: (i) −*x*+2, −*y*+2, −*z*+1.

Fig. 1

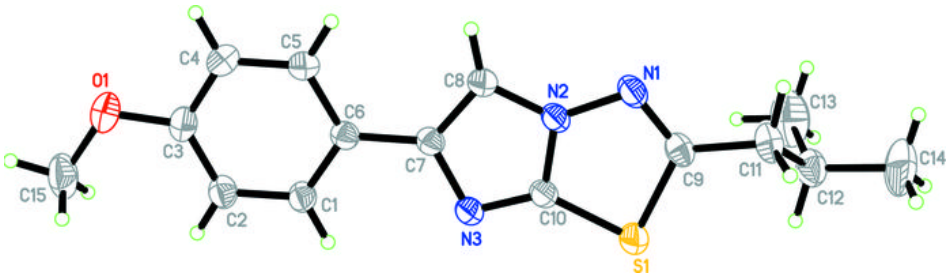


Fig. 2